Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Lei Cao,‡ Jia-Geng Liu and Duan-Jun Xu\*

Department of Chemistry, Zhejiang University, Hangzhou, People's Republic of China

‡ also at Institute of Geological Survey of Jiangsu Province, Nanjing, People's Republic of China

Correspondence e-mail: xudj@mail.hz.zj.cn

#### Key indicators

Single-crystal X-ray study T = 295 KMean  $\sigma(\text{C-C}) = 0.005 \text{ Å}$  R factor = 0.054 wR factor = 0.104 Data-to-parameter ratio = 17.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography All rights reserved

# Tris(1*H*-benzimidazole- $\kappa N^3$ )(thiodiacetato- $\kappa^3 O, S, O'$ )cobalt(II) dihydrate

In the title complex,  $[Co(C_4H_4O_4S)(C_7H_6N_2)_3] \cdot 2H_2O$ , the Co<sup>II</sup> ion is coordinated by one thiodiacetate dianion and three benzimidazole molecules, with a distorted octahedral geometry. The tridentate thiodiacetate chelates to the Co atom in a facial configuration, with a Co-S bond distance of 2.5597 (8) Å. Intermolecular hydrogen bonding prevents the face-to-face stacking of neighboring benzimdazole ligands.

### Comment

Several crystal structures of metal complexes incorporating benzimidazole ( $C_7H_6N_2$ ; bzim) as a ligand show  $\pi$ - $\pi$  stacking between neighboring aromatic rings (Chen *et al.*, 2003; Liu & Xu, 2004; Bukowska-Strzyzewska & Tosik, 1983). As part of our ongoing investigations on the nature of  $\pi$ - $\pi$  stacking in metal complexes, (Li *et al.*, 2005), the title bzim complex of Co<sup>II</sup>, (I) (Fig. 1), is presented here.



The Co<sup>II</sup> atom is coordinated by three bzim molecules and one tridentate thiodiacetate dianion (C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>S<sup>2-</sup>; tda) in a distorted octahedral coordination (Table 1). The tda ligand adopts a facial configuration, consistent with that found in triimidazole(thiodiacetato)nickel(II) (Pan & Xu, 2005). The five-membered chelate rings of the tda display envelope conformations, with the Co atom lying at the flap position, displaced from the mean planes formed by the other four atoms by 0.370 (5) and 0.836 (4) Å for the O1 and O3 rings, respectively. The Co–S bond distance in (I) is slightly longer [by 0.0848 (11) Å] than the equivalent Ni–S bond distance found in the above-cited Ni<sup>II</sup> complex.

Extensive hydrogen bonding (Table 2) occurs in the crystal structure of (I). As shown in the packing diagram (Fig. 2), the uncoordinated tda carboxyl O atoms link with the bzim ligands of neighboring complexes *via*  $N-H\cdots$ O hydrogen bonding, preventing any close  $\pi-\pi$  stacking (minimum centroid separation = 4.43 Å) of neighboring bzim ligands in the crystal structure of (I).

Received 8 February 2006 Accepted 16 February 2006

## **Experimental**

An aqueous solution (15 ml) solution of  $CoCl_2 \cdot 6H_2O$  (1 mmol),  $H_2$ tda (1 mmol) and  $Na_2CO_3$  (1 mmol) was mixed with an ethanol solution (5 ml) of bzim (2 mmol). The solution was refluxed for 6 h and then filtered. Single crystals of (I) were obtained after 6 d.

Mo  $K\alpha$  radiation Cell parameters from 28828

reflections

 $\begin{array}{l} \theta = 1.9 \mbox{--} 27.4^{\circ} \\ \mu = 0.78 \ \mbox{mm}^{-1} \end{array}$ 

T = 295 (2) K

 $0.24 \times 0.20 \times 0.06 \ \text{mm}$ 

Plate, red

### Crystal data

$[C_0(C_4H_4O_4S)(C_7H_6N_2)_3]\cdot 2H_2C_3$
$M_r = 597.51$
Orthorhombic, Pbca
a = 10.3428 (4) Å
b = 19.8783 (4) Å
c = 25.8749 (8) Å
V = 5319.8 (3) Å <sup>3</sup>
Z = 8
$D_x = 1.492 \text{ Mg m}^{-3}$

### Data collection

Rigaku R-AXIS RAPID	6053 independent reflections
diffractometer	3934 reflections with $I > 2\sigma(I)$
$\omega$ scans	$R_{\rm int} = 0.095$
Absorption correction: multi-scan	$\theta_{\rm max} = 27.5^{\circ}$
ABSCOR (Higashi, 1995)	$h = -13 \rightarrow 13$
$T_{\min} = 0.888, T_{\max} = 0.955$	$k = -25 \rightarrow 24$
46075 measured reflections	$l = -33 \rightarrow 33$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0379P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.054$	+ 3.8108P]
$wR(F^2) = 0.104$	where $P = (F_{o}^{2} + 2F_{c}^{2})/3$
S = 1.05	$(\Delta/\sigma)_{\rm max} = 0.001$
6053 reflections	$\Delta \rho_{\rm max} = 0.37 \text{ e } \text{\AA}^{-3}$
352 parameters	$\Delta \rho_{\rm min} = -0.36 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

#### Table 1

Selected bond lengths (Å).

Co-S	2.5597 (8)	Co-N13	2.154 (3)
Co-O1	2.056 (2)	Co-N23	2.109 (2)
Co-O3	2.156 (2)	Co-N33	2.136 (2)

Table 1	2
---------	---

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$O5-H5A\cdots O6^{i}$	0.95	2.04	2.870 (4)	145
$O5-H5B\cdots O4$	0.96	1.85	2.805 (4)	177
$O6-H6A\cdots O2$	0.94	1.93	2.769 (4)	148
$O6-H6B\cdots O5^{ii}$	0.96	1.88	2.816 (4)	162
N11-H11···O4 <sup>iii</sup>	0.86	1.93	2.766 (4)	163
$N21 - H21 \cdot \cdot \cdot O3^{iv}$	0.86	2.39	3.217 (3)	161
$N21 - H21 \cdots O4^{iv}$	0.86	2.42	3.072 (3)	133
$N31 - H31 \cdots O2^{v}$	0.86	1.93	2.780 (4)	168

Symmetry codes: (i) -x + 1,  $y + \frac{1}{2}$ ,  $-z + \frac{3}{2}$ ; (ii)  $-x + \frac{1}{2}$ ,  $y - \frac{1}{2}$ , z; (iii) x + 1, y, z; (iv)  $x + \frac{1}{2}$ ,  $-y + \frac{3}{2}$ , -z + 1; (v)  $-x + \frac{3}{2}$ ,  $y + \frac{1}{2}$ , z.

Water H atoms were located in a difference Fourier map and refined as riding in their as-found relative positions with fixed  $U_{iso}(H) = 0.05 \text{ Å}^2$ . Other H atoms were placed in idealized positions (C-H = 0.93–0.97; N-H = 0.86 Å) and refined as riding, with  $U_{iso}(H) = 1.2U_{eq}(\text{carrier})$ .

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/



### Figure 1

The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). Dashed lines indicate hydrogen bonds.



#### Figure 2

The packing of (I), showing  $N-H\cdots O$  hydrogen bonds as dashed lines. Non-coordinated water molecules and H atoms have been omitted for clarity.

MSC and Rigaku, 2002); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

The work was supported by the National Natural Science Foundation of China (grant No. 20443003).

### References

- Altomare, A., Cascarano, G., Giacovazzo, C. & Guagliardi, A. (1993). J. Appl. Cryst. 26, 343–350.
- Bukowska-Strzyzewska, M. & Tosik, A. (1983). Acta Cryst. C39, 203-205.

Chen, Z., Xu, D.-J., Li, Z.-Y., Wu, J.-Y. & Chiang, M. Y. (2003). J. Coord. Chem. 56, 253–259.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.
- Li, H., Yin, K.-L. & Xu, D.-J. (2005). Acta Cryst. C61, m19-m21.
- Liu, Y. & Xu, D.-J. (2004). Acta Cryst. E60, m1002-m1004.
- Pan, T.-T. & Xu, D.-J. (2005). Acta Cryst. E61, m1735-1737.
- Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSC and Rigaku (2002). CrystalStructure. Version 3.00. Rigaku/MSC, 900 New Trails Drive, The Woodlands, TX 77381-5209, USA.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.